

SUBJECT: SCIENCE
SYLLABUS BREAKUP FOR THE ACADEMIC SESSION: 2026-27
CLASS: X
SUBJECT: BIOLOGY

S.NO.	MONTH	CHAPTER	PRACTICAL
1.	APRIL	Chapter 5: Life Processes Nutrition in Plants and animals.	Preparing a temporary mount of a leaf peel to show stomata.
	I WEEKLY TEST TERM I 04/05/2026	Chapter 5: Life Processes Nutrition in Plants and Amoeba	
2.	MAY	Chapter 5: Life Processes Respiration, Transportation, Excretion.	Experimentally show that carbon dioxide is given out during respiration.
3.	JULY	Chapter 6: Control and Coordination	
	II WEEKLY TEST TERM I 20/07/2026	Chapter 5: Life Processes Nutrition in Humans, Respiration, Transportation, Excretion.	
4.	AUGUST	Chapter 7: How do Organisms Reproduce	Studying (a) binary fission in Amoeba, and (b) budding in yeast and Hydra with the help of prepared slides
5.	SEPTEMBER	Chapter 8: Heredity	Identification of the different parts of an embryo of a dicot seed (Pea, gram or red kidney bean).
6.	SEPTEMBER	Chapter 13: Our Environment	
	REVISION TEST OCTOBER	WHOLE SYLLABUS	
	I PRE-BOARD NOVEMBER	WHOLE SYLLABUS	
	II PRE-BOARD DECEMBER	WHOLE SYLLABUS	

SUBJECT: PHYSICS

S.NO.	MONTH	CHAPTER	NO. OF PERIODS
1.	APRIL-MAY	Chapter 9: Light: Reflection and Refraction <ul style="list-style-type: none"> ● 9.1 Reflection of light ● 9.2 Spherical mirrors ● Image formation by spherical mirror ● Sign convention for reflection by spherical mirrors ● Mirror formula and magnification ● Numerical NCERT Questions and Answers discussion	TOTAL = 11 1 2 2 1 1 3 1
	I WEEKLY TEST TERM I 04/05/2026	CHAPTER 9: LIGHT - REFLECTION (up to Spherical mirrors and numerical based on mirror formula)	
2.	MAY	Chapter 9: Light: Reflection and Refraction <ul style="list-style-type: none"> ● 9.3 Refraction of light ● 9.3.1 Refraction through a rectangular glass slab ● 9.3.2 Refractive index ● 9.3.3 Refraction by spherical lenses ● 9.3.4 Image formation by lenses ● 9.3.6 Sign convention for spherical lenses ● 9.3.7 Lens formula and magnification ● 9.3.8 Power of a lens ● Numerical 	TOTAL =13 1 1 2 2 1 1 1 2 1

		<ul style="list-style-type: none"> ● Practical applications of heating effect of electric current ● 11.8 Electric power ● Interrelation between P, V, I and R Numerical NCERT Questions and Answers discussion	2 1
5.	SEPTEMBER	CHAPTER 12: MAGNETIC EFFECTS OF ELECTRIC CURRENT <ul style="list-style-type: none"> ● 12.1 Magnetic field and field lines ● 12.2 Magnetic field due to a current carrying conductor ● 12.2.1 Magnetic field due to current through a straight conductor ● 12.2.2 Right hand thumb rule ● 12.2.3 Magnetic field due to current through a circular loop. ● 12.2.4 Magnetic field due to current in a solenoid. ● 12.3 Force on a current carrying conductor in a magnetic field. ● Fleming's Left Hand Rule ● Direct current, Alternating current, frequency of AC. ● Advantage of AC over DC ● 12.4 Domestic electric circuits ● NCERT Questions and Answers discussion *EXTRA TOPICS Motor, Electromagnetic Induction, Electric Generator	TOTAL =11 1 1 1 1 1 1 1 1 1 1
	OCTOBER	REVISION TEST	10
	I PRE-BOARD EXAMINATION NOVEMBER 2026		
	II PRE-BOARD EXAMINATION DECEMBER 2026		

THE WHOLE SYLLABUS WILL BE EVALUATED IN I & II PRE-BOARD EXAMINATION

LIST OF EXPERIMENTS IN PHYSICS

1. To determine of the focal length of a given concave mirror.
2. To determine of the focal length of a given convex lens.
3. To trace the path of a ray of light passing through a rectangular glass slab for different angles of incidence. Measure the angle of incidence, angle of refraction, angle of emergence and interpret the result.
4. To trace the path of the rays of light through a glass prism.
5. To study the dependence of potential difference (V) across a resistor on the current (I) passing through it and determine its resistance. Also plotting a graph between V and I.
6. To determine of the equivalent resistance of two resistors when connected in series and parallel.

SUBJECT: CHEMISTRY

S.NO.	MONTH	CHAPTER	
1.	APRIL-MAY	CHAPTER 1 CHEMICAL REACTIONS AND EQUATIONS	TOTAL = 6
		1.1 Chemical equations <ul style="list-style-type: none"> ● Balancing by hit trial method. 	1
		1.2 Types of chemical reactions. <ul style="list-style-type: none"> ● Combination reaction ● Decomposition reaction ● Single displacement reaction ● Double displacement reaction ● Oxidation – reduction reaction 	3
		1.3 Corrosion and rancidity	1
		NCERT question and answer discussion.	1
	I WEEKLY TEST TERM I 04/05/2026	CHAPTER 1 (CHEMICALS REACTIONS AND EQUATIONS TILL TOPIC 1.2.2 DECOMPOSITION REACTION)	
2.	MAY-JULY	CHAPTER 2: ACIDS, BASES AND SALTS.	TOTAL = 8
		2.1 Chemical properties of acids and bases. <ul style="list-style-type: none"> ● Chemical properties of acids. ● Chemical properties of bases. 	2
		2.2 What do all acids and bases have in common? <ul style="list-style-type: none"> ● Dilution of an acid. ● Dilution of a base. 	1

		<p>2.3 pH of different solutions.</p> <ul style="list-style-type: none"> ● Importance of pH in daily life. <p>2.4 Salts (Definition, classification and Properties)</p> <ul style="list-style-type: none"> ● Common salt. ● Sodium hydroxide ● Bleaching powder. ● Baking soda. ● Washing soda. ● Plaster of paris <p>NCERT question and answer discussion.</p>	<p>1</p> <p>3</p> <p>1</p>
	<p>II WEEKLY TEST TERM I 20/07/2026</p>	<p>CHAPTER 1: (CHEMICALS REACTIONS AND EQUATIONS FROM TOPIC 1.2.3 TO 1.2.5) CHAPTER 2: (ACIDS, BASES AND SALTS TILL TOPIC 2.1.6)</p>	
3.	<p>JULY-AUGUST</p>	<p>CHAPTER 3: METAL AND NON METALS</p> <p>3.1 Physical properties.</p> <ul style="list-style-type: none"> ● Metals. ● Non metals. <p>3.2 Chemical properties of metals.</p> <ul style="list-style-type: none"> ● Reaction in air. ● Reaction with water. ● Reaction with acids. ● Reaction with the solution of other metals. <p>3.3 reaction of metal and non metals.</p> <ul style="list-style-type: none"> ● Formation of ionic compounds. ● Properties of ionic compounds. <p>3.4 Extraction of metals.</p> <ul style="list-style-type: none"> ● Enrichment of ores. ● Extraction of metals on the basis of activity series. ● Refining of metals. <p>3.5 Corrosion.</p> <p>NCERT question and answer discussion</p>	<p>TOTAL = 8</p> <p>1</p> <p>2</p> <p>1</p> <p>2</p> <p>1</p> <p>1</p>
4.	<p>AUGUST-SEPTEMBER</p>	<p>CHAPTER 4: CARBON AND ITS COMPOUNDS</p> <p>4.1 The covalent bond.</p> <ul style="list-style-type: none"> ● Single covalent bond. 	<p>TOTAL = 11</p>

	<ul style="list-style-type: none"> ● Double covalent bond. ● Triple covalent bond. <p>4.2 Versatile nature of Carbon.</p> <ul style="list-style-type: none"> ● Tetravalency. ● Catenation. ● Formation of multiple bonds. ● Isomerism. ● Homologous series. ● IUPAC nomenclature of carbon compounds <ul style="list-style-type: none"> ○ Hydrocarbons (Alkane, alkene and alkyne). ○ Alcohols Carbonyl compounds (Aldehyde and ketons). ○ Carboxylic acids. <p>4.3 Chemical properties of carbon compounds.</p> <ul style="list-style-type: none"> ● Combustion reaction. ● Oxidation reaction. ● Addition reaction. ● Substitution reaction. <p>4.4 Some important carbon compounds.</p> <ul style="list-style-type: none"> ● Ethanol (Physical and chemical properties). ● Ethanoic acid (Physical and chemical properties). <p>4.5 Soaps and detergents</p> <ul style="list-style-type: none"> ● Introduction. ● Cleansing action of soaps. <p>NCERT question and answer discussion</p>	<p>1</p> <p>4</p> <p>2</p> <p>2</p> <p>1</p> <p>1</p>
	I PRE-BOARD EXAMINATION NOVEMBER 2026	
	II PRE-BOARD EXAMINATION DECEMBER 2026	
	THE WHOLE SYLLABUS WILL BE EVALUATED IN I & II PRE-BOARD EXAMINATION	

LIST OF EXPERIMENTS IN CHEMISTRY

FIRST TERM:

Experiment 1: To find the pH of the following samples using pH paper/ universal indicator.

(2 Periods)

- a) Dilute hydrochloric acid.
- b) Dilute NaOH solution.
- c) Dilute ethanoic acid solution.
- d) Lemon juice.
- e) Water.

f) Dilute sodium bicarbonate solution.

Experiment 2: To study the properties of acids and bases (Dil. HCl and Dil. NaOH) by their reaction
(2 Periods)

- a) Litmus solution (Blue/ Red).
- b) Zinc Metal.
- c) Solid sodium carbonate.

Experiment 3: To perform and observe the following reaction and classify them into:
(2 Periods)

- a) Action of water on quicklime (Combination reaction)
- b) Action of heat on ferrous sulphate solution (Decomposition reaction)
- c) Iron nails kept in solution of copper sulphate (Displacement reaction)
- d) Reaction between sodium sulphate and barium chloride solutions (double displacement reaction).

Experiment 4: To observe the action of Zn, Fe, Cu and Al metals on the following salts solutions:
(2 Periods)

(a) $ZnSO_4$ (b) $FeSO_4$ (c) $CuSO_4$ (d) $Al_2(SO_4)_3$

And arrange Zn, Fe, Cu and Al metals in increasing order of reactivity based on the above result.

SECOND TERM:

Experiment 1: To study the following properties of acetic acid (ethanoic acid):
(2 Periods)

- a) Odour
- b) Solubility in water
- c) Effect on litmus solution
- d) Reaction with solid sodium carbonate.

Experiment 2: To study the saponification reaction for the preparation of soap.
(2 Periods)

Experiment 3: To compare the foaming capacity of different samples of soap.
(2 Periods)
